

**N-(3-Bromophenyl)phthalimide****Kong Mun Lo and Seik Weng Ng\***Department of Chemistry, University of Malaya,  
50603 Kuala Lumpur, Malaysia

Correspondence e-mail: seikweng@um.edu.my

**Key indicators**

Single-crystal X-ray study

 $T = 298\text{ K}$ Mean  $\sigma(\text{C-C}) = 0.005\text{ \AA}$  $R$  factor = 0.036 $wR$  factor = 0.075

Data-to-parameter ratio = 12.5

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

The benzene and phthalimido portions of the molecule of *N*-(3-bromophenyl)phthalimide,  $\text{C}_{14}\text{H}_8\text{BrNO}_2$ , are twisted by  $38.1(1)\text{ \AA}$ ; two molecules are linked across a center of inversion by  $\text{Br}\cdots\text{O}$  interactions of length  $3.093(3)\text{ \AA}$ .

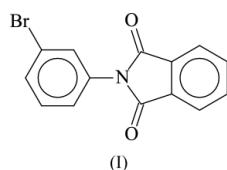
Received 19 April 2004

Accepted 20 April 2004

Online 24 April 2004

**Comment**

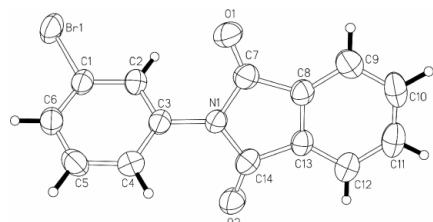
In the crystal structure of *N*-(2-bromophenyl)phthalimide, (I), the benzene and phthalimido portions of the molecule are almost perpendicular to each other; the structure features an intermolecular  $\text{Br}\cdots\text{O}$  contact of about  $3.1\text{ \AA}$  that is interpreted in terms of possible charge-transfer or dipole–dipole interactions (Wu *et al.*, 2002). The two portions are twisted by  $38.1(1)^\circ$  in the present 3-bromo analog (Fig. 1), which also features a similar intermolecular interaction of  $3.093(3)\text{ \AA}$  (Fig. 2).

**Experimental**

Phthalic anhydride (5 g, 30 mmol) and 3-bromoaniline (3.7 ml, 30 mmol) were heated in ethanol (100 ml) for 2 h. The solution was filtered; slow cooling of the solution yielded the impure compound, which was then purified by recrystallization from ethanol (m.p. 421–423 K).

**Crystal data**

$\text{C}_{14}\text{H}_8\text{BrNO}_2$	$D_x = 1.712\text{ Mg m}^{-3}$
$M_r = 302.12$	Mo $\text{K}\alpha$ radiation
Monoclinic, $P2_1/n$	Cell parameters from 25 reflections
$a = 12.025(1)\text{ \AA}$	$\theta = 5.4\text{--}12.5^\circ$
$b = 7.877(1)\text{ \AA}$	$\mu = 3.50\text{ mm}^{-1}$
$c = 12.877(1)\text{ \AA}$	$T = 298(2)\text{ K}$
$\beta = 106.03(1)^\circ$	Block, colorless
$V = 1172.3(1)\text{ \AA}^3$	$0.4 \times 0.4 \times 0.3\text{ mm}$
$Z = 4$	

**Figure 1**

ORTEPII (Johnson, 1976) plot of (I), with displacement ellipsoids drawn at the 50% probability level. H atoms are shown as spheres of arbitrary radii.

**Data collection**

Enraf–Nonius CAD-4 diffractometer  
 $\omega$ – $\theta$  scans  
 Absorption correction: empirical via  $\psi$  scan (North *et al.*, 1968)  
 $T_{\min} = 0.269$ ,  $T_{\max} = 0.350$   
 2160 measured reflections  
 2054 independent reflections  
 1433 reflections with  $I > 2\sigma(I)$

**Refinement**

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.036$   
 $wR(F^2) = 0.075$   
 $S = 0.99$   
 2054 reflections  
 164 parameters  
 H-atom parameters constrained

$R_{\text{int}} = 0.020$   
 $\theta_{\text{max}} = 25.0^\circ$   
 $h = 0 \rightarrow 14$   
 $k = -9 \rightarrow 0$   
 $l = -15 \rightarrow 14$   
 3 standard reflections frequency: 60 min  
 intensity decay: none

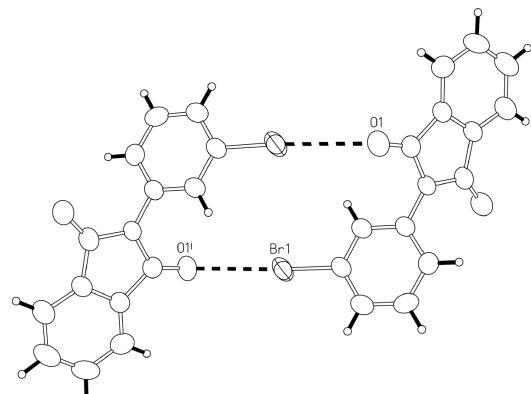
$$w = 1/[\sigma^2(F_o^2) + (0.0304P)^2 + 0.8453P]$$

where  $P = (F_o^2 + 2F_c^2)/3$

 $(\Delta/\sigma)_{\text{max}} = 0.001$   
 $\Delta\rho_{\text{max}} = 0.27 \text{ e } \text{\AA}^{-3}$   
 $\Delta\rho_{\text{min}} = -0.32 \text{ e } \text{\AA}^{-3}$   
 Extinction correction: *SHELXL97*  
 Extinction coefficient: 0.0194 (10)
**Table 1**Selected geometric parameters ( $\text{\AA}$ ,  $^\circ$ ).

Br1–C1	1.899 (4)	C4–C5	1.387 (5)
O1–C7	1.196 (4)	C5–C6	1.375 (5)
O2–C14	1.201 (4)	C7–C8	1.489 (5)
N1–C14	1.400 (4)	C8–C13	1.377 (5)
N1–C7	1.411 (4)	C8–C9	1.377 (5)
N1–C3	1.433 (4)	C9–C10	1.387 (5)
C1–C6	1.365 (5)	C10–C11	1.372 (6)
C1–C2	1.380 (4)	C11–C12	1.388 (5)
C2–C3	1.375 (5)	C12–C13	1.379 (5)
C3–C4	1.376 (5)	C13–C14	1.486 (5)
C14–N1–C3	124.9 (3)	N1–C7–C8	105.0 (3)
C14–N1–C7	112.3 (3)	C7–C8–C9	129.9 (4)
C3–N1–C7	122.8 (3)	C7–C8–C13	108.5 (3)
C2–C1–C6	121.8 (3)	C9–C8–C13	121.6 (3)
C6–C1–Br1	119.3 (3)	C8–C9–C10	117.2 (4)
C2–C1–Br1	118.9 (3)	C9–C10–C11	121.1 (4)
C1–C2–C3	118.4 (3)	C10–C11–C12	121.7 (4)
C2–C3–C4	121.3 (3)	C11–C12–C13	116.8 (4)
C2–C3–N1	118.9 (3)	C12–C13–C8	121.6 (3)
C4–C3–N1	119.8 (3)	C12–C13–C14	129.4 (3)
C3–C4–C5	118.7 (3)	C8–C13–C14	109.0 (3)
C4–C5–C6	120.9 (4)	O2–C14–N1	125.5 (3)
C1–C6–C5	118.9 (3)	O2–C14–C13	129.4 (3)
O1–C7–N1	125.5 (3)	N1–C14–C13	105.1 (3)
O1–C7–C8	129.5 (3)		

H atoms were placed at calculated positions [ $\text{C}–\text{H} = 0.93 \text{ \AA}$  and  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ ] and were refined using a riding model.

**Figure 2**

ORTEPII (Johnson, 1976) plot of (I), illustrating the  $\text{Br}\cdots\text{O}$  interactions as dashed lines [symmetry code: (i)  $1 - x, -y, 1 - z$ ].

Data collection: *CAD-4/PC* (Kretschmar, 1994); cell refinement: *CAD-4 VAX/PC Fortran System* (Enraf–Nonius, 1988); data reduction: *XCAD4* (Harms, 1997) in *WinGX* (Farrugia, 1999); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEPII* (Johnson, 1976); software used to prepare material for publication: *SHELXL97*.

The authors thank the Ministry of Science, Technology & the Environment (IRPA 09-02-03-0100 EA100) and the University of Malaya for supporting this work.

**References**

- Enraf–Nonius (1988). *CAD-4 VAX/PC Fortran System*. Operator's Guide to the Enraf–Nonius CAD-4 Diffractometer Hardware, its Software and the Operating System. Enraf–Nonius, Scientific Instruments Division, PO Box 483, 2600 AL Delft, The Netherlands.
- Farrugia, L. J. (1999). *J. Appl. Cryst.* **32**, 837–838.
- Harms, K. (1997). *XCAD4*. University of Marburg, Germany.
- Johnson, C. K. (1976). *ORTEPII*. Report ORNL-5138. Oak Ridge National Laboratory, Tennessee, USA.
- Kretschmar, M. (1994). *CAD-4/PC*. Version 1.5c. University of Tübingen, Germany.
- North, A. C. T., Phillips, D. C. & Mathews, F. S. (1968). *Acta Cryst.* **A24**, 351–359.
- Sheldrick, G. M. (1997). *SHELXL97* and *SHELXS97*. University of Göttingen, Germany.
- Wu, J.-Y., Chiang, M. Y.-N. & Zeng, W.-F. (2002). *Acta Cryst.* **E58**, o1370–o1371.